5 SEM TDC DSE CHM (CBCS) 2 (H)

2021

(Held in January/February, 2022)

CHEMISTRY

(Discipline Specific Elective)

(For Honours)

Paper: DSE-2

(Green Chemistry)

Full Marks: 53
Pass Marks: 21

Time: 3 hours

The figures in the margin indicate full marks for the questions

1. Choose the correct answer:

1×6=6

- (a) The Bhopal Disaster (1984) was occurred due to
 - (i) methyl cyanide
 - (ii) methyl isocyanide
 - (iii) methyl isocvanate
 - (iv) methyl cyanate
- (b) The Japanese disease Itai-Itai has been attributed to
 - (i) lead poisoning
 - (ii) mercury poisoning
 - (iii) cadmium poisoning
 - (iv) arsenic poisoning

- (c) 12 green chemistry principles are postulated by
 - (i) Professor Paul T. Anastas
 - (ii) Professor John C. Warner
 - (iii) Professor Paul T. Anastas and Professor John C. Warner
 - (iv) Professor John R. Asthana
- (d) The formula of adipic acid is
 - (i) $HOOC-(CH_2)_2-COOH$
 - (ii) HOOC—(CH₂)₆—COOH
 - (iii) HOOC—(CH2)4—COOH
 - (iv) $HOOC-(CH_2)_3-COOH$
- (e) Which of the following reactions is an example of microwave-assisted reaction in water?
 - (i) Hoffmann elimination
 - (ii) Oxidation of toluene
 - (iii) Oxidation of alcohol
 - (iv) All of the above
 - (f) An efficient, green synthesis of a compostable and widely applicable plastic made from corn is
 - (i) polylactic acid
 - (ii) polyacetic acid
 - (iii) polyvinyl chloride
 - (iv) polyacrylic acid

2. Answer any ten of the following questions:

2×10=20

- (a) Explain the term 'green chemistry'.
- (b) What are the goals of green chemistry?
- (c) Define atom economy.
- (d) How can you improve the atom economy of a reaction?
- (e) How can you compare the greenness of solvents?
- (f) What is chemoselective reaction? Give one example of it.
- (g) What is enantioselective reaction? Give one example of it.
- (h) Write the reactions involved during the depletion of ozone layer by CFCs.
- (i) Write the green approach of synthesis of adipic acid.
- (j) Write the alternative approach to Strecker synthesis for the synthesis of disodium iminodiacetate (DSIDA).
- (k) Write the greener approach to the Bhopal Gas Tragedy.
- (l) What are solid-state reactions? Give one example.
- 3. Write short notes on any three of the following: $3\times3=9$
 - (a) Principles of green chemistry
 - (b) Ionic liquids as green solvents

- (c) supercritical carbon dioxide
- (d) Protection of a functional group
- **4.** Answer any *three* of the following questions:

3×3=9

- (a) Compare the oxidation of toluene and oxidation of alcohols using microwave conditions in water with other conventional procedures.
- (b) Compare the Simmons-Smith reaction using ultrasound conditions with other conventional procedures.
- (c) How can we design the environmentally safe marine antifoulant?
- (d) Write a note on surfactants for carbon dioxide.
- 5. Answer any three of the following questions:

3×3~

- (a) What will be the future trends in green chemistry in the field of oxidative transformations and catalysis?
- (b) "Green chemistry is sustainable chemistry." Explain the statement.
- (c) What will be the future trends in green chemistry in the field of biominimetic, multifunctional reagents?
- (d) What will be the future trends in green chemistry in the field of proliferation of solventless reactions?
